



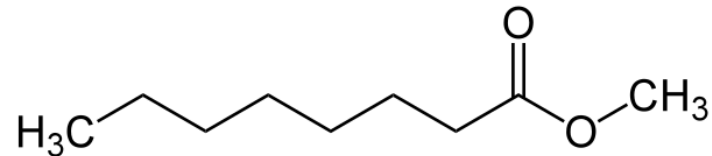
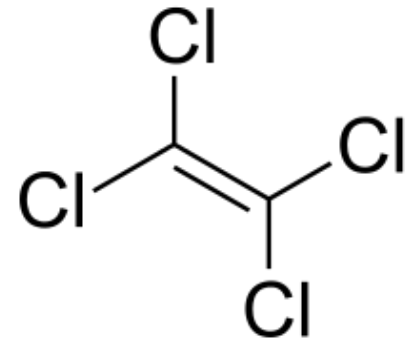
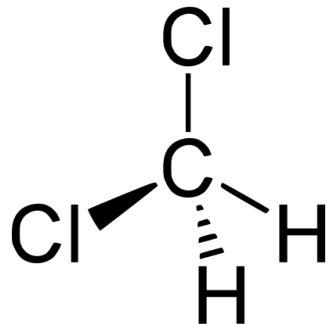
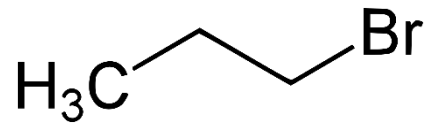
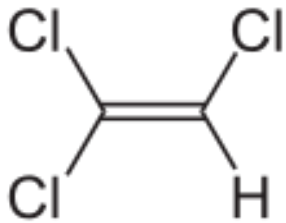
ALWAYS  
2 STEPS  
AHEAD ...



# Development testing equipment

Matthias Martus

infraTest Prüftechnik GmbH



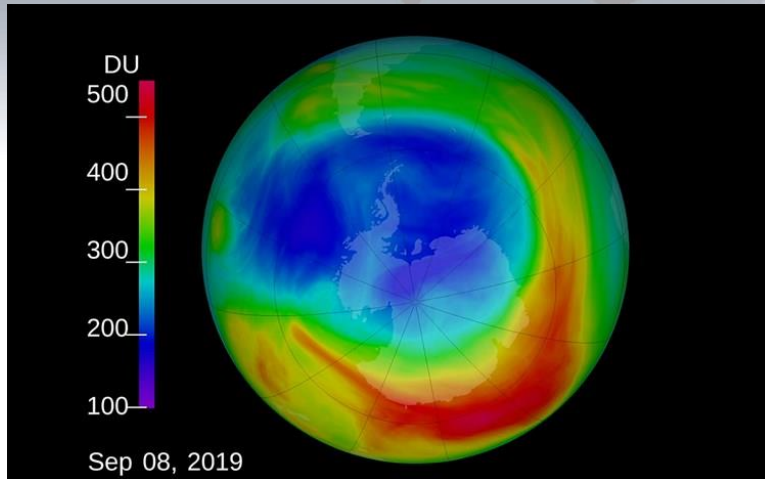
# Solvents in Asphalt Extraction

# REACH

Registration,  
Evaluation,  
Authorisation and  
Restriction of **Chemicals**

 **ECHA**  
EUROPEAN CHEMICALS AGENCY

# Yes solvents are not nice: Thats why they are solvents



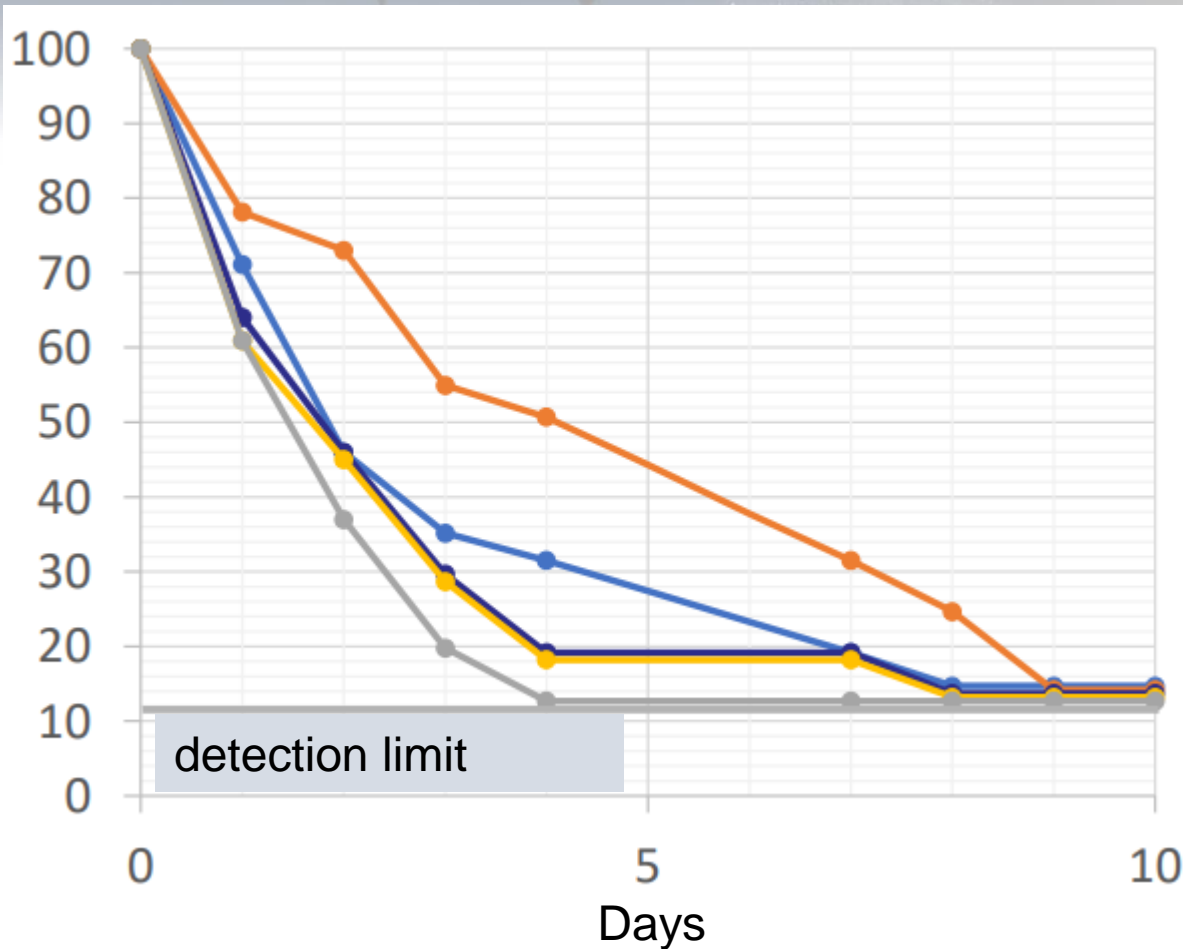
# Some key facts

Solvents in asphalt extraction are in direct contact with aggregates, binder, polymers, waxes, recycled materials and unknown materials



# Example: Boiling test Aggregate/Solvent:

Concentration of Amino based Stabilizer



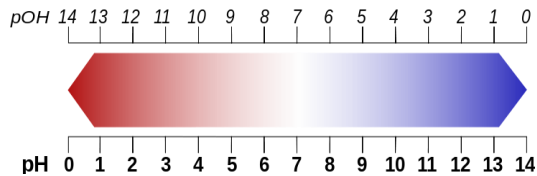
Aggregates:

- Diabase
- Greywacke

With different grading curve and therefore different dimension of surface

# Some key facts

Solvents have to be handled as a test fluid and must be monitored and tested regularly



pH value and alkalinity in ppm have to be determined

	<a href="#">Batteriesäure</a>	<0	sour
	<a href="#">Magensäure</a> (nüchterner Magen)	1,0–1,5	
	<a href="#">Zitronensaft</a>	2,4	
	<a href="#">Cola</a>	2,0–3,0	
	<a href="#">Essig</a>	2,5	
	Fruchtsaft der <a href="#">Schattenmorelle</a>	2,7	
	<a href="#">Wein</a>	4,0	
	Saure <a href="#">Milch</a>	4,5	
	<a href="#">Bier</a>	4,5–5,0	
	<a href="#">Saurer Regen</a>	< 5,0	
	<a href="#">Kaffee</a>	5,0	
	<a href="#">Hautoberfläche</a> des Menschen	5,5	
	<a href="#">Regen</a> (natürlicher Niederschlag)	5,6	
	<a href="#">Mineralwasser</a>	6,0	
	<a href="#">Milch</a>	6,5	
	<a href="#">Wasser</a> (je nach <a href="#">Härte</a> )	6,0–8,5	
	Menschlicher <a href="#">Speichel</a>	6,5–7,4	alkaline
	<a href="#">Blut</a>	7,4	
	<a href="#">Meerwasser</a>	7,5–8,4	
	<a href="#">Pankreassaft (Darmsaft)</a>	8,3	
	<a href="#">Seife</a>	9,0–10,0	
	Haushalts- <a href="#">Ammoniak</a>	11,5	
	<a href="#">Bleichmittel</a>	12,5	
	<a href="#">Beton</a>	12,6	
	<a href="#">Natronlauge</a>	13,5–14	



# Some key facts

- Solvents in use should have a purity of higher 99,5 %
- Solvent mixtures have different influence on binder
- Solvent mixtures have most likely different boiling points
- All recycled solvents will destroy the machine due to unknown substances



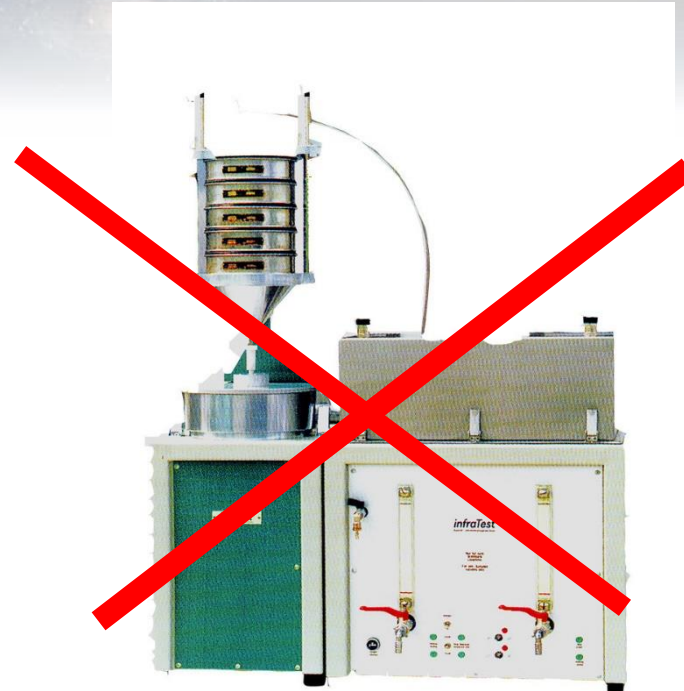
# Important Solvent questions

- 1. What is the possible impact on human health ?**
- 2. What is the possible environmental impact ?**
- 3. What is the influence on binder property ?**
- 4. How is the dissolving power (extraction time)**
- 5. How do we get it out of the binder ?**
- 6. How stable is it ?**
- 7. Is it flammable ?**
- 8. Is it explosive ?**
- 9. Is it available ?**
- 10. Is it affordable ?**
- 11. How can we recycle it ?**

# Answer to 1 and 2: Closed systems



**Asphaltanalysator**

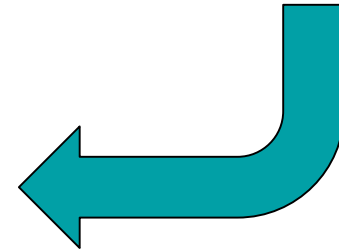
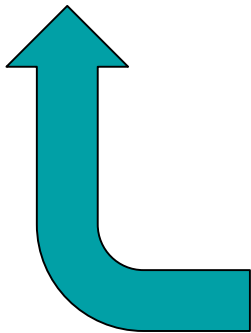


**Automatic Extractor**

# Answer to 1 and 2: Closed systems



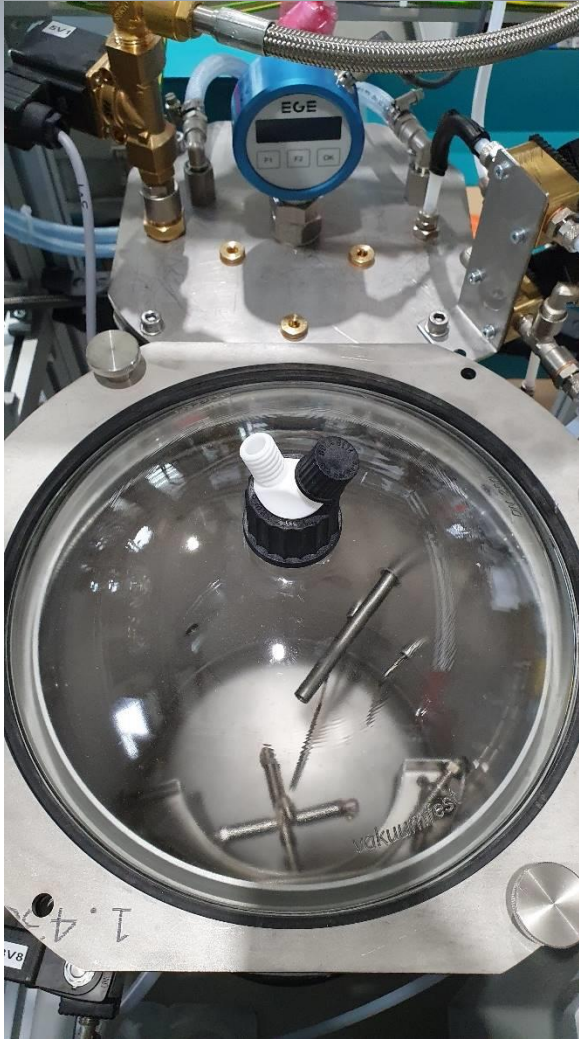
# Answer to 1 and 2: Closed systems



# Answer to 1 and 2: Closed systems



# Closed system indirect recovery



# Answer to 1 and 2: Closed systems





# Answer to 1 and 2

**1.1 What is the possible impact on human health ?**

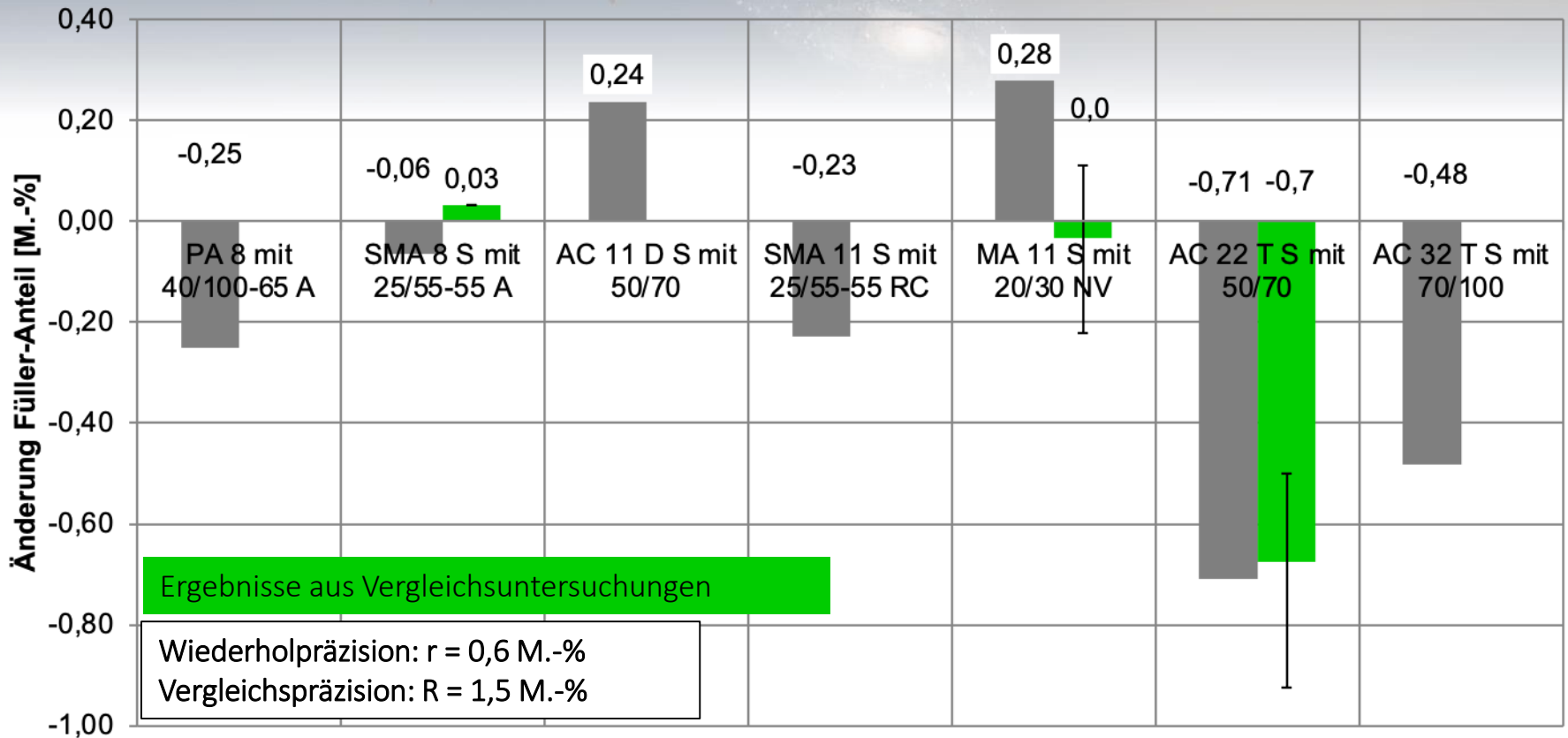
**1.2 What is the possible environmental impact ?**



# Answer to 1 and 2: Closed systems

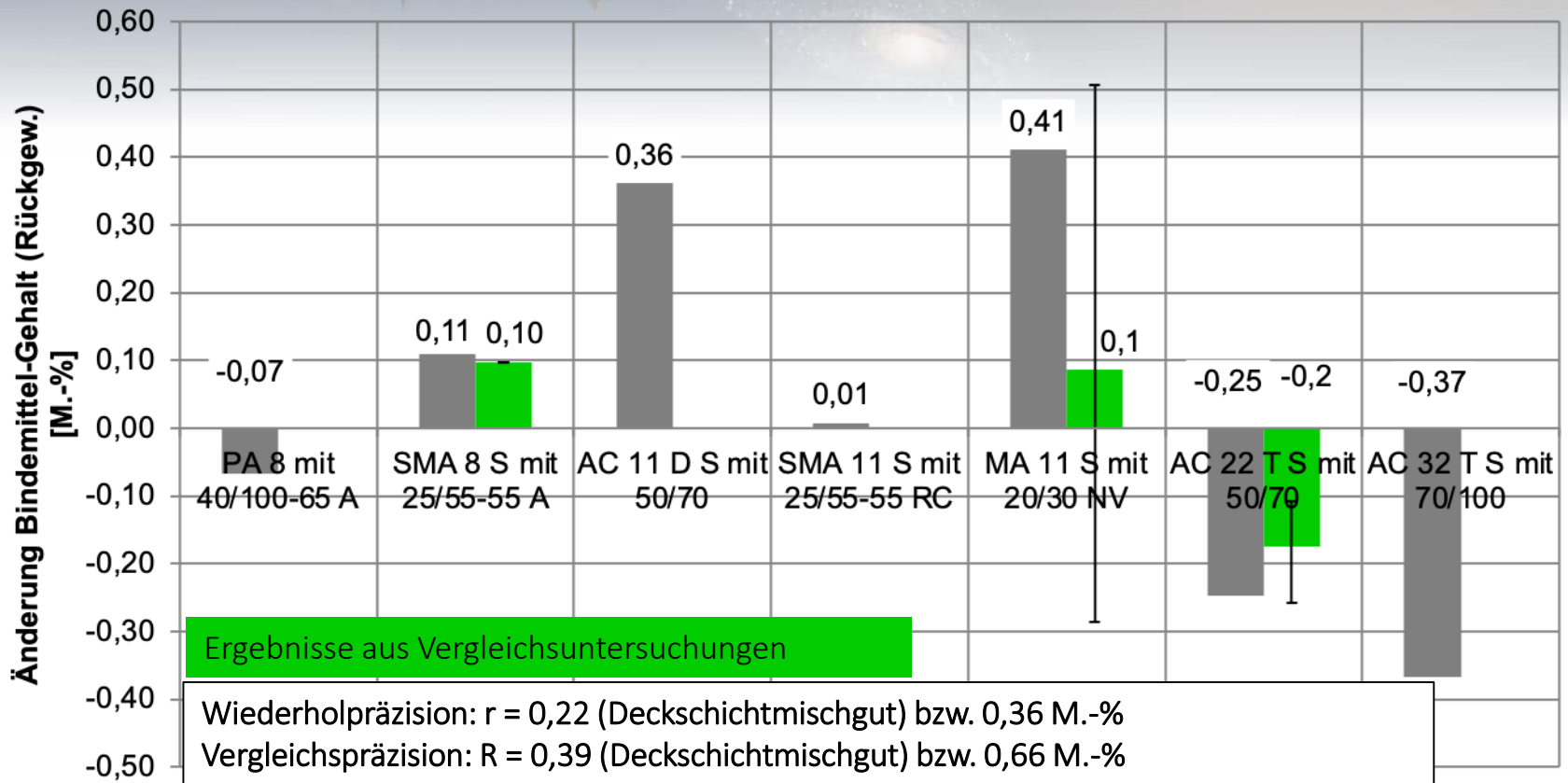
- **Safety belt and Airbag = internal, machine based**
- **Traffic rules and signs = external, system based**
- **E and hydrogen cars**
  
- **Conclusion: Make a known machine and system as safe as you can**

# 3. Filler content OME versus TRI



→ Die Bestimmung des Füllergehalts mit OME ist gleichermaßen präzise wie mit Tri

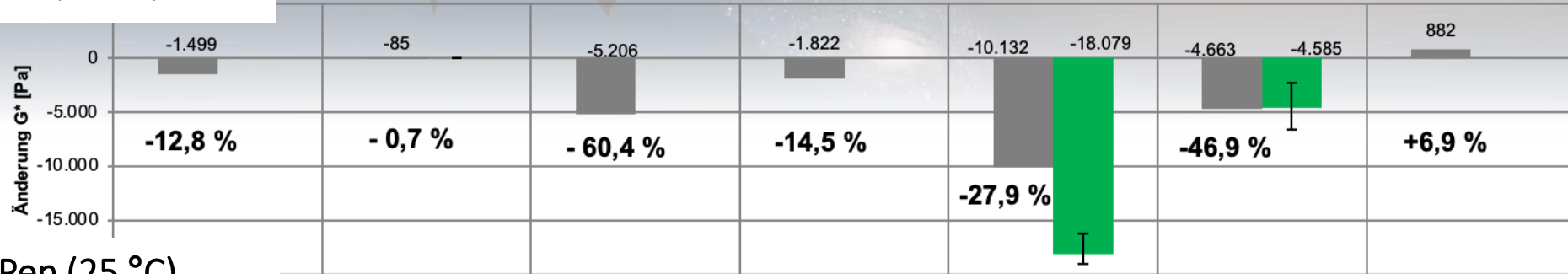
# 3. Binder Content OME versus TRI



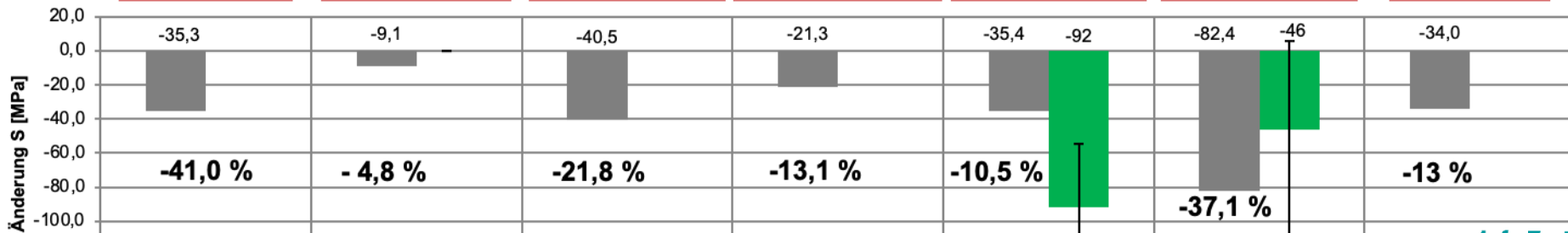
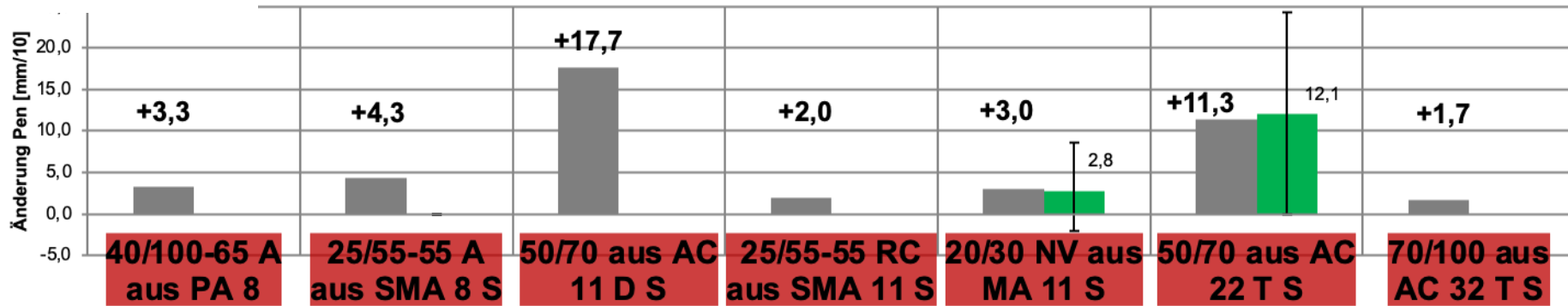
→ Die Bestimmung des Bindemittelgehalts mit OME ist gleichermaßen präzise wie mit Tri

# 3. Binder properties OME versus TRI

G\* (60 °C)

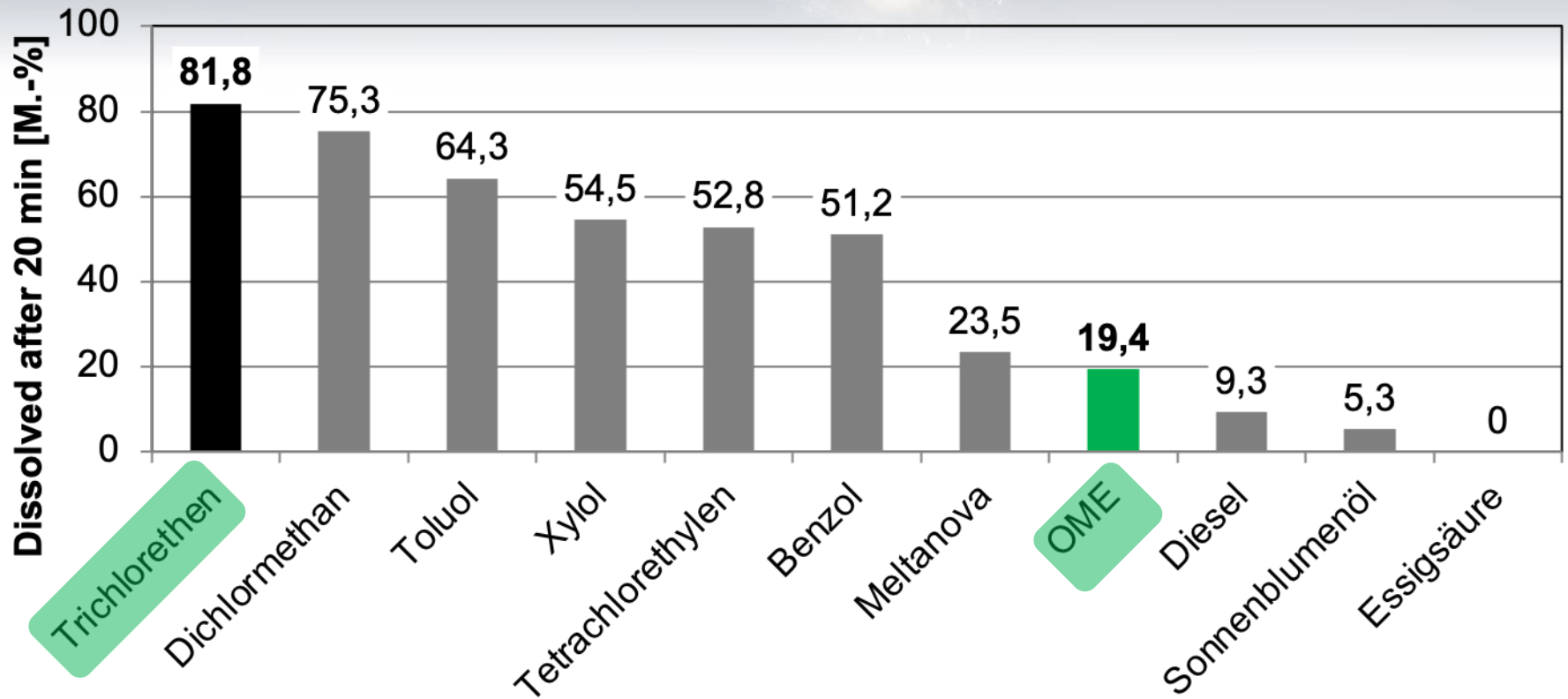


Pen (25 °C)

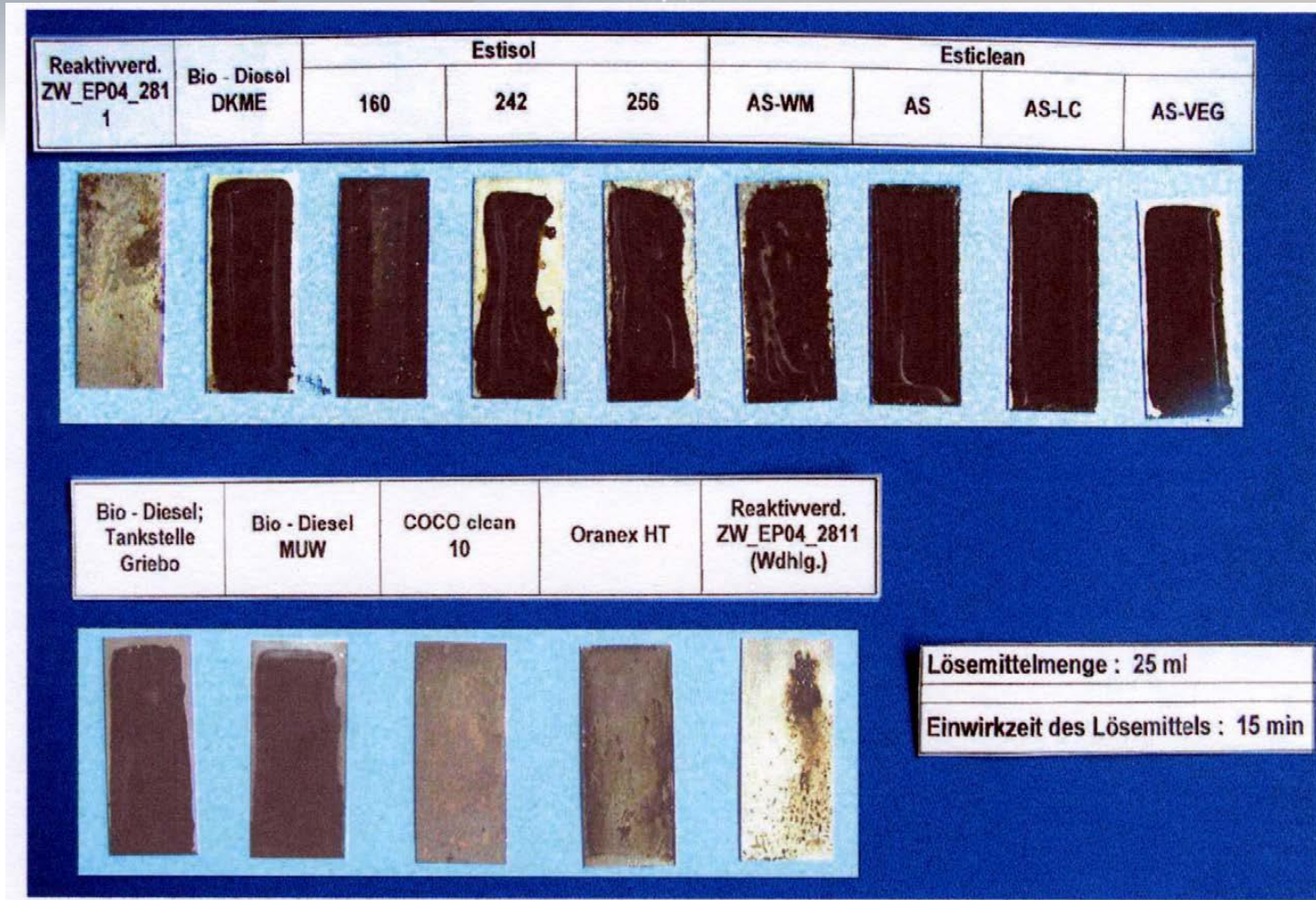


S (-16 °C)

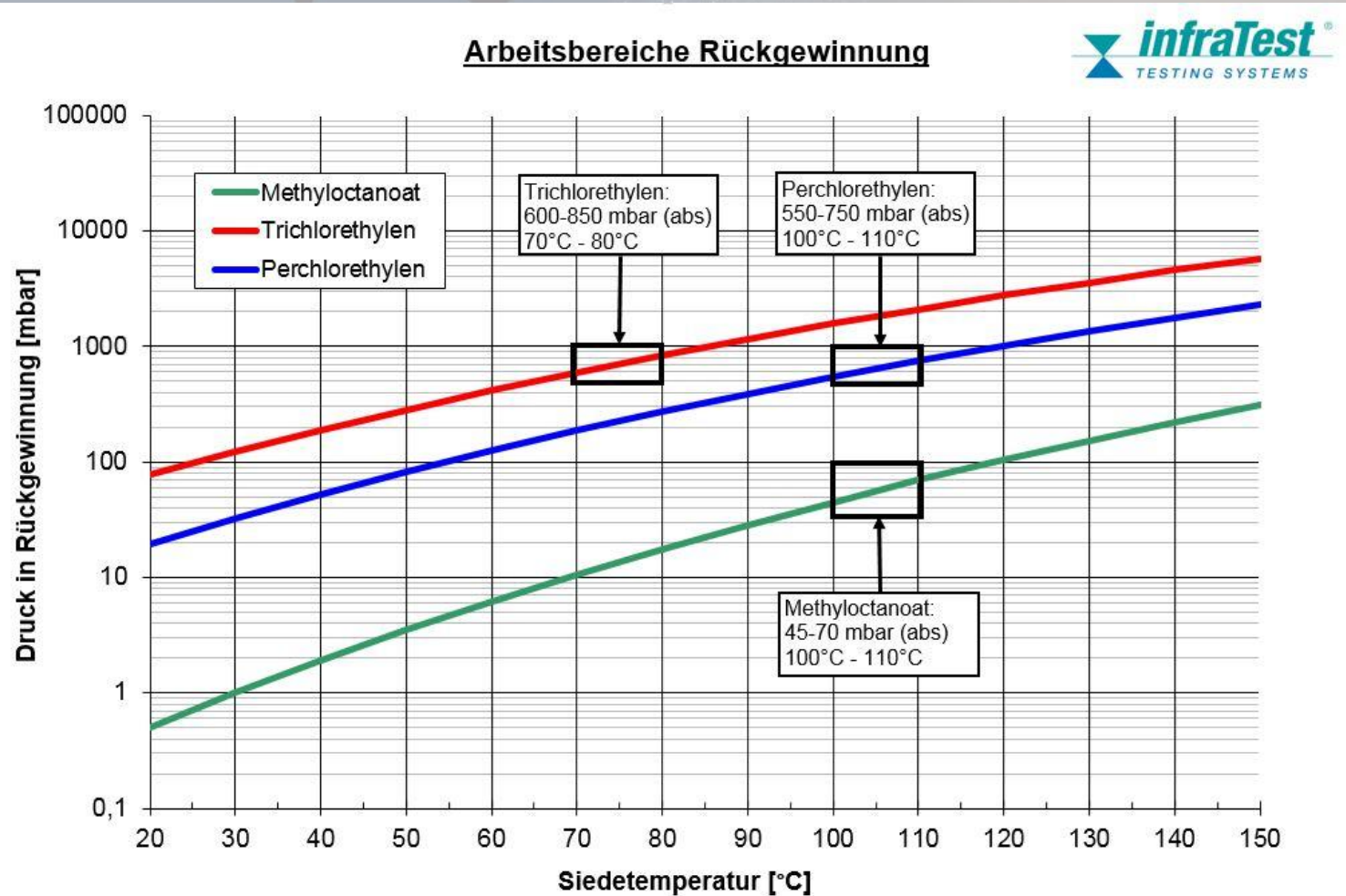
# 4. Dissolving power of different solvents at 26°C



# 4. Dissolving power of different solvents at 26°C

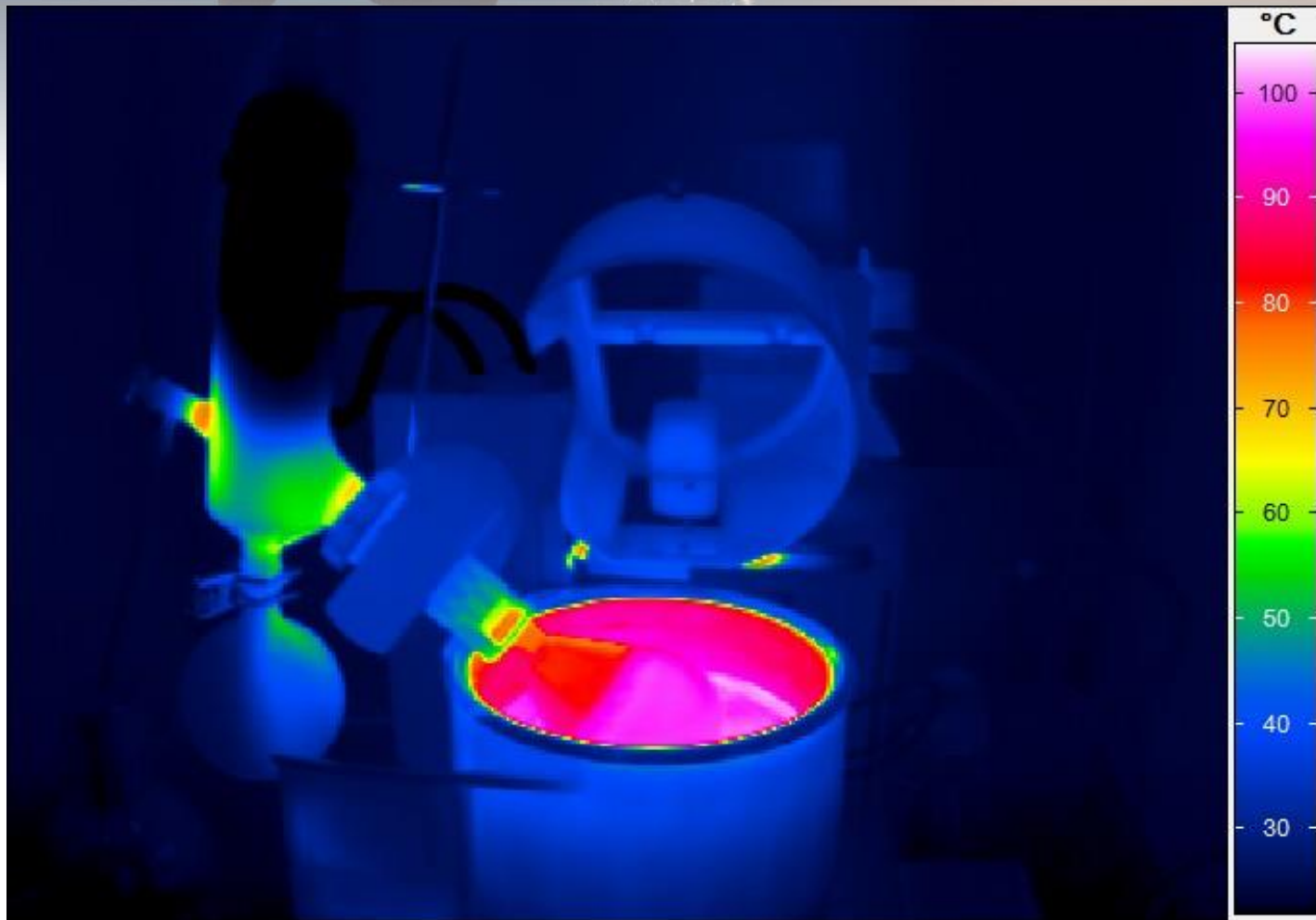


# 5. Distillation: Boiling Temperature and Pressure

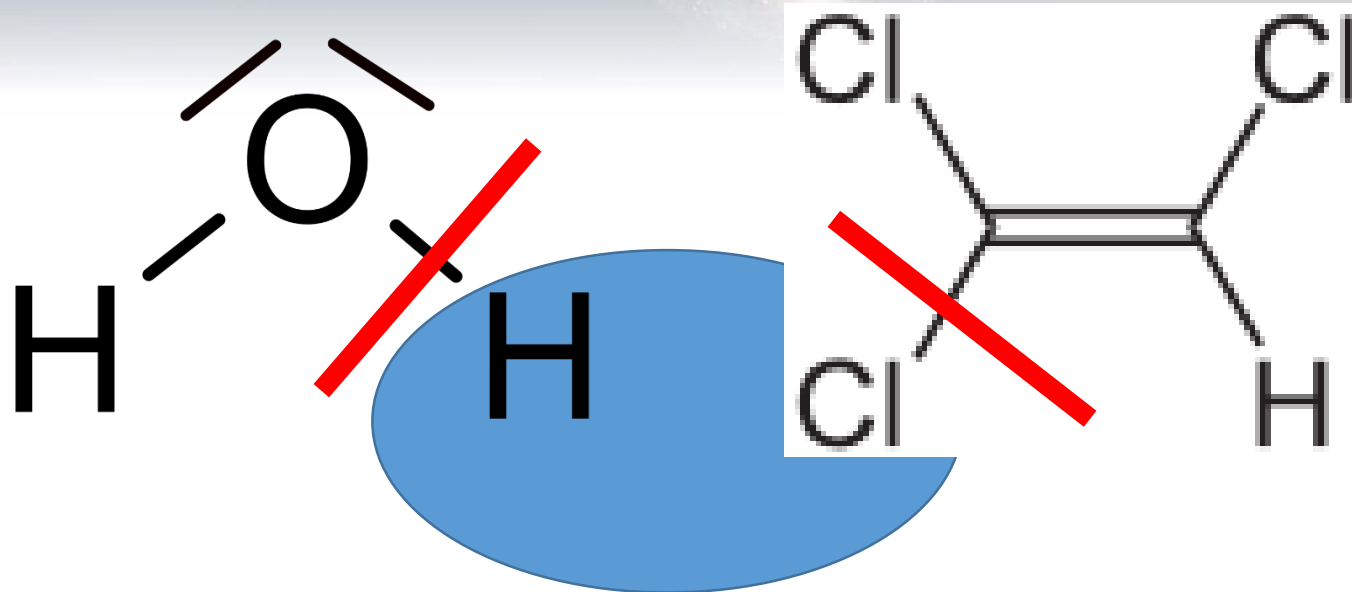




# 5. Temperature level at Vaccum Evaporator



## 6. How stable is the solvent during use ?



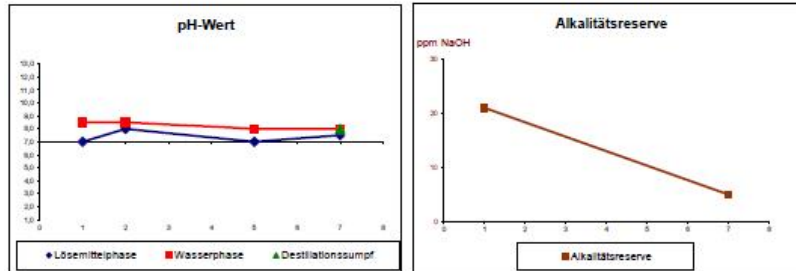
= HCL (hydrochloric acid)

## 6. How stable is the solvent during use ?

- 1. Influence of aggregats binders and polymers etc.**
- 2. Solvent is ageing**
- 3. Unstable against UV and heat**
- 4. Certain metals ( for ex. from milling machine)**
- 5. Rust in buckets**
- 6. Malfunction Analyser**

# 6. How stable is the solvent during use ?

## Analysedetails



### Lösemittelphase

pH-Wert  
Alkalitätsreserve  
Chlorid-Test  
Kommentar

Tag	1	2	3	4	5	6	7	8
pH-Wert	7,0	8,0			7,0		7,5	
Alkalitätsreserve	21,0						5,0	
Chlorid-Test	neg.	neg.			neg.		neg.	
Kommentar	Klar und farblos							

### Wasserphase

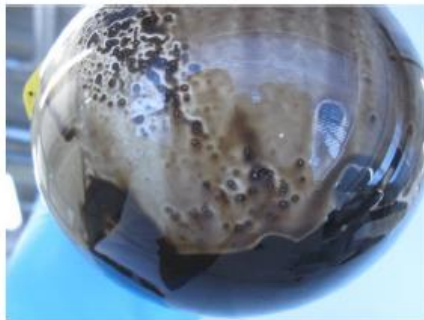
pH-Wert  
Chlorid-Test  
Kommentar

Tag	1	2	3	4	5	6	7	8
pH-Wert	8,5	8,5			8,0		8,0	
Chlorid-Test	neg.	neg.			neg.		neg.	

### Destillationsumpf

pH-Wert  
Kommentar

Tag	1	2	3	4	5	6	7	8
pH-Wert							8,0	
Kommentar	trübe dunkelbraune Flüssigkeit, mit schwarzen Ablagerungen und "Steinpartikeln", graue Verkrustungen am Thermometer.							



## Metalle

Aluminium  
Eisen  
Messing  
Kupfer

ok  
ok  
leichte Verfärbung  
Verfärbung

Metalle/ Metals/ Metaux/ Metali/ Metales

Aluminium Aluminio	Unbenutzt Unused Inutilizado Inutilizzato Inutilizzato Nach dem Test After the test Après le test Dopo il test Después de la prueba	
Eisen Iron Fer Ferro Hierro	Unbenutzt Unused Inutilizado Inutilizzato Inutilizzato Nach dem Test After the test Après le test Dopo il test Después de la prueba	
Messing Brass Laiton Ottone Latón	Unbenutzt Unused Inutilizado Inutilizzato Inutilizzato Nach dem Test After the test Après le test Dopo il test Después de la prueba	
Edestahl Stainless steel Acier inoxydable Acciaio inossidabile Acero fino	Unbenutzt Unused Inutilizado Inutilizzato Inutilizzato Nach dem Test After the test Après le test Dopo il test Después de la prueba	
Kupfer Copper Cobre Rame Cobre	Unbenutzt Unused Inutilizado Inutilizzato Inutilizzato Nach dem Test After the test Après le test Dopo il test Después de la prueba	

SA-Nr.: 09772-00909

Datum: 09.03.17      Signatur:

# 7 and 8: flammables and explosives



Currently no machine is designed for that kind of solvents

Pricing is 150 % higher then now

# 9 and 10: available, affordable ?

**Solvents are usually produced in distillation columns like this.**

**If you need 100 t per year for the asphalt extraction nobody will care**



# 9 and 10: available, affordable ?

The project coconutester (methyl octanoate) also died because nobody wanted to produce or deliver it( and we dont have enough coconuts in germany and sweden I guess)



# 11: recycle ?

Solvent handling  
always must  
include  
a procedure  
for recycling





# 11: recycle ?

Chemical leasing concepts are a possible solution



# Which solvents are in use

1.1 Trichlorethylen

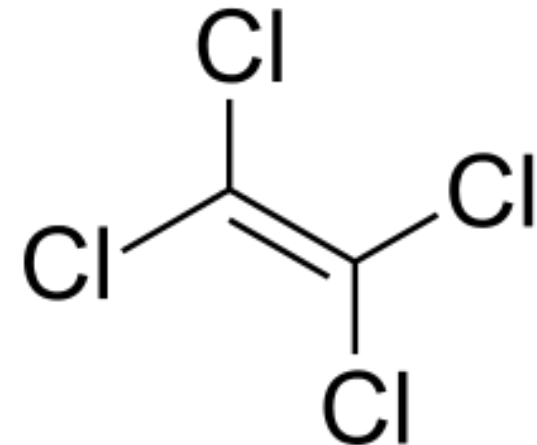
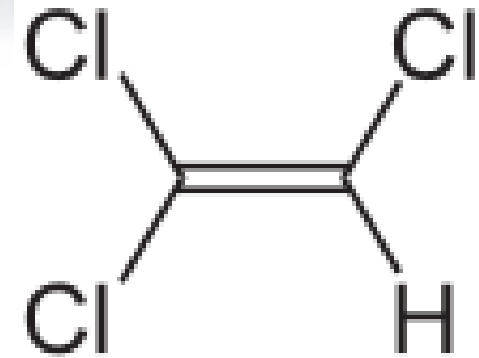
1.2 Tetrachlorethylen

1.3 Methylenchlorid

1.4 Toluene

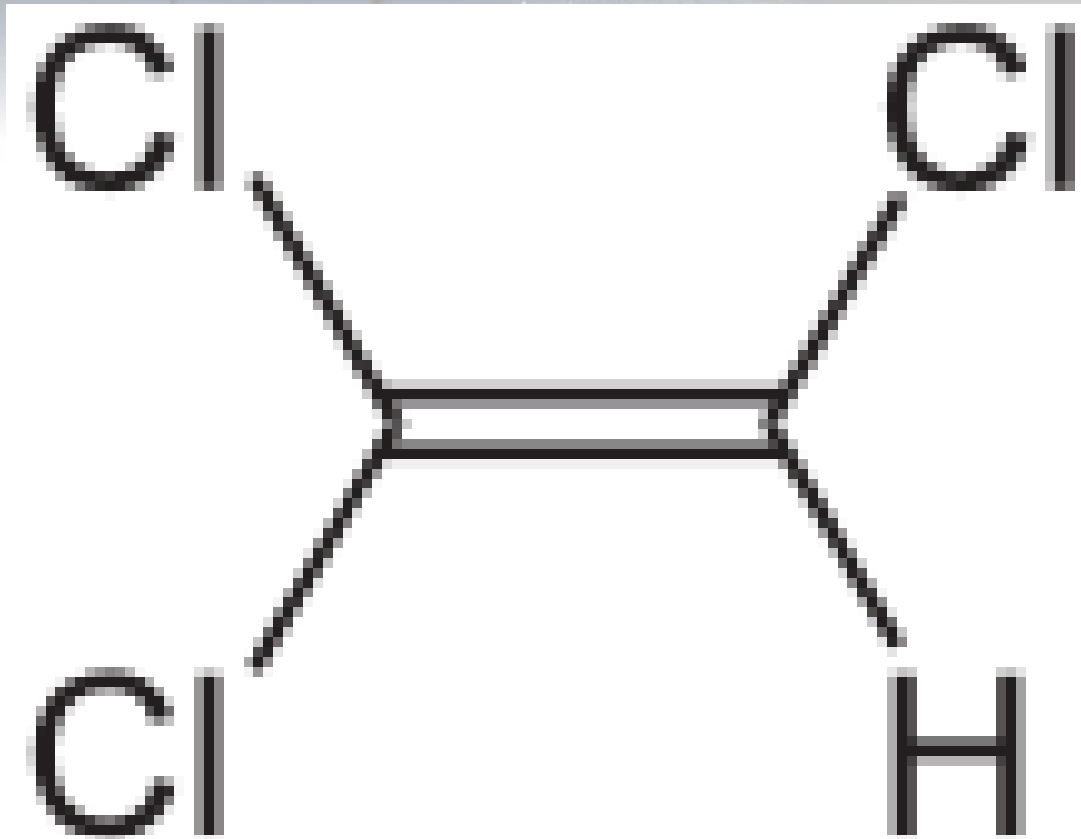
1.5 Methyl Octanoate

1.6 n-propyl bromide



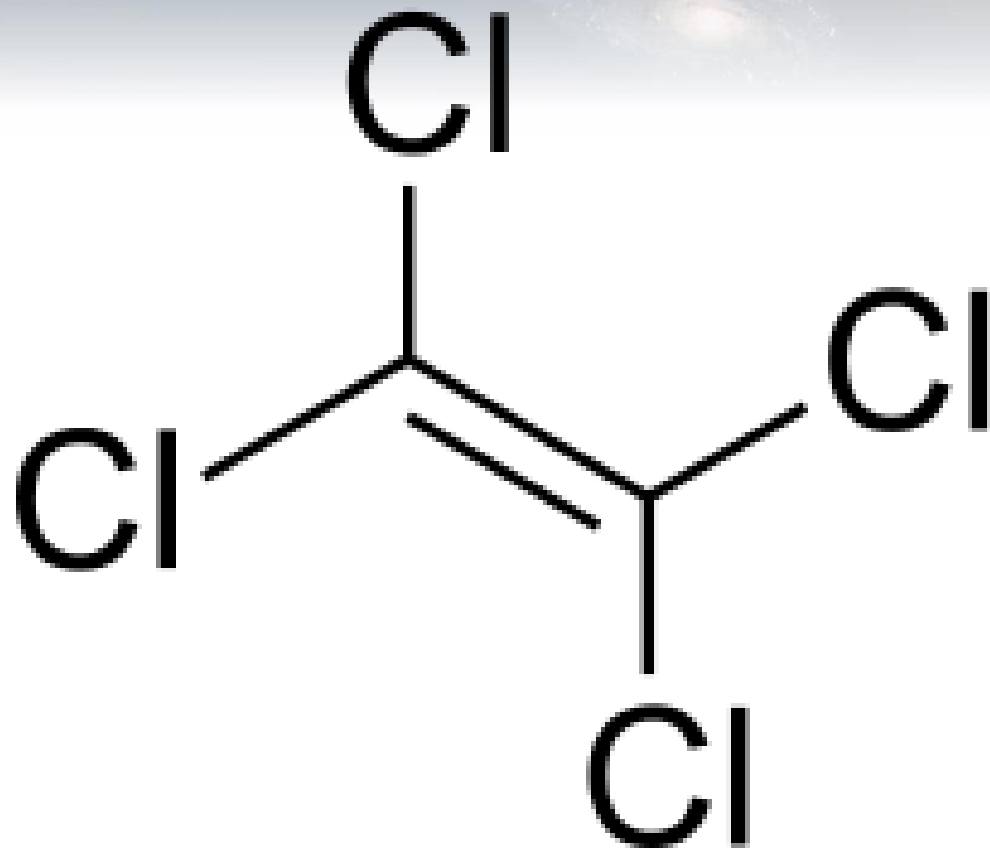
## 1.1 What is TRI ?

- Chlorinated hydrocarbon
- Density: 1,46 g /cm<sup>3</sup>
- Boiling point: 87 ° C
- SVHC ( permission for use will expire 04/23)



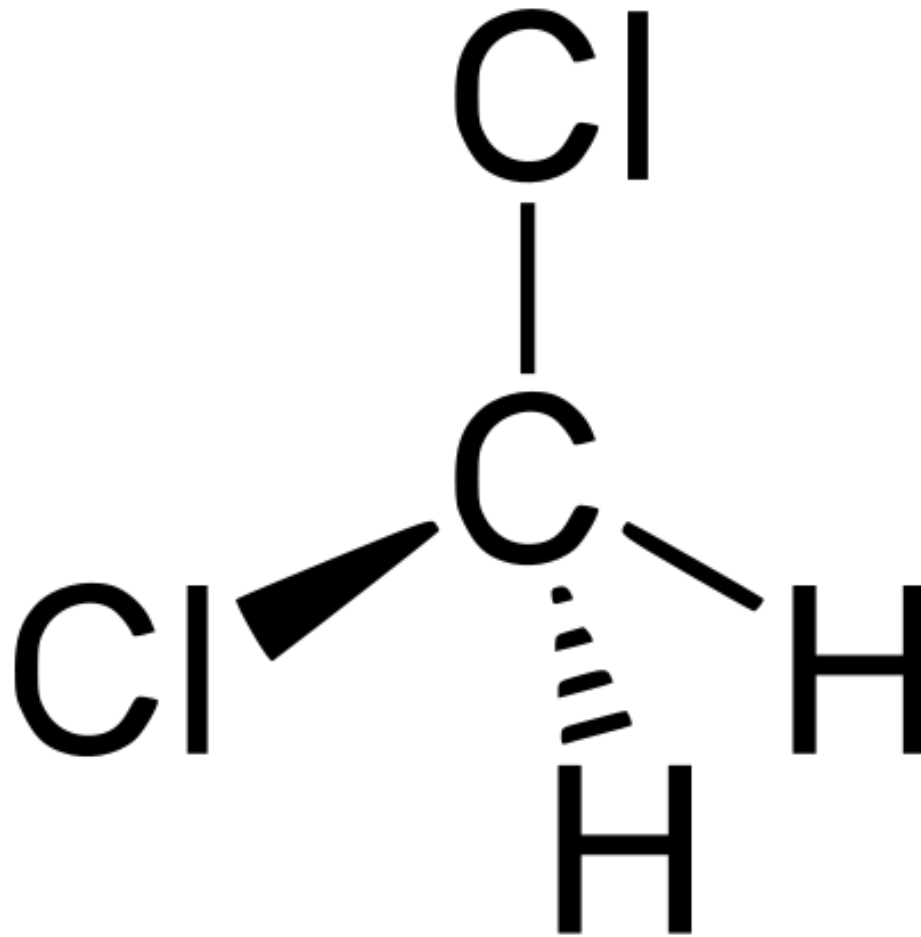
## 1.2 What is Tetra ?

- Chlorinated hydrocarbon
- Density: 1,62 g /cm<sup>3</sup>
- Boiling point: 121,1 ° C
- not listed as a SVHC



## 1.2 What is Methylenechloride?

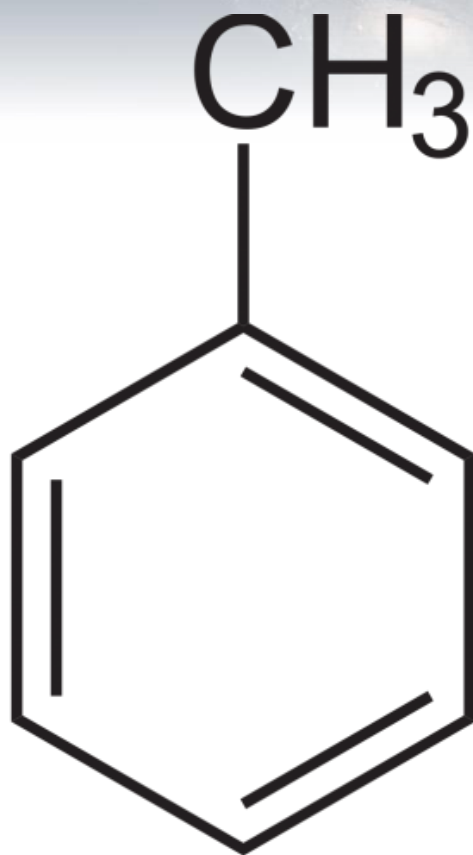
- Chlorinated hydrocarbon
- Density: 1,33 g /cm<sup>3</sup>
- Boiling point: 40 ° C
- Not listed as a SVHC





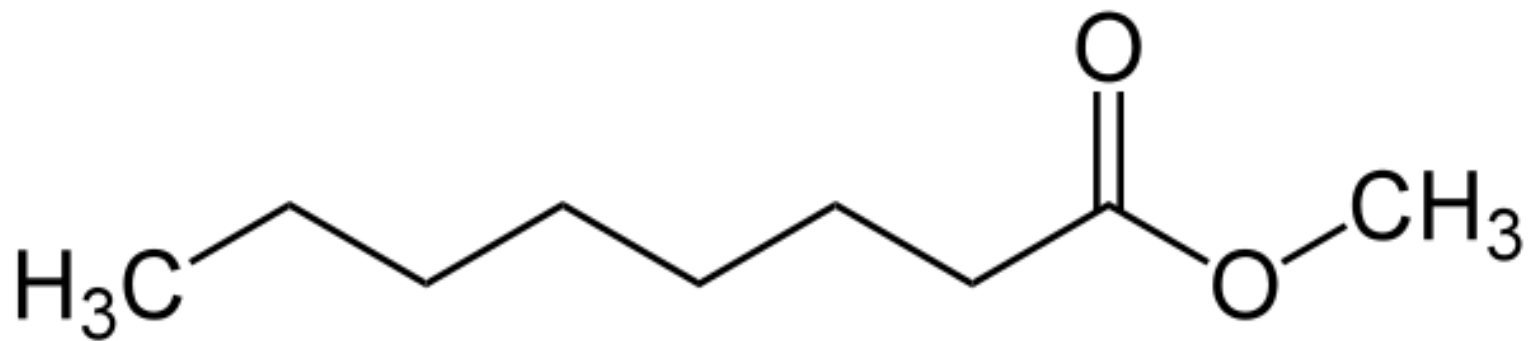
## 1.2 What is Toluene?

- Aromatic hydrocarbon
- Density: 0,87 g /cm<sup>3</sup>
- Boiling point: 108 ° C
- Lower and upper explosion limit:  
1,1 vol% to 7,1 vol%
- Flash point: 4,4 ° C



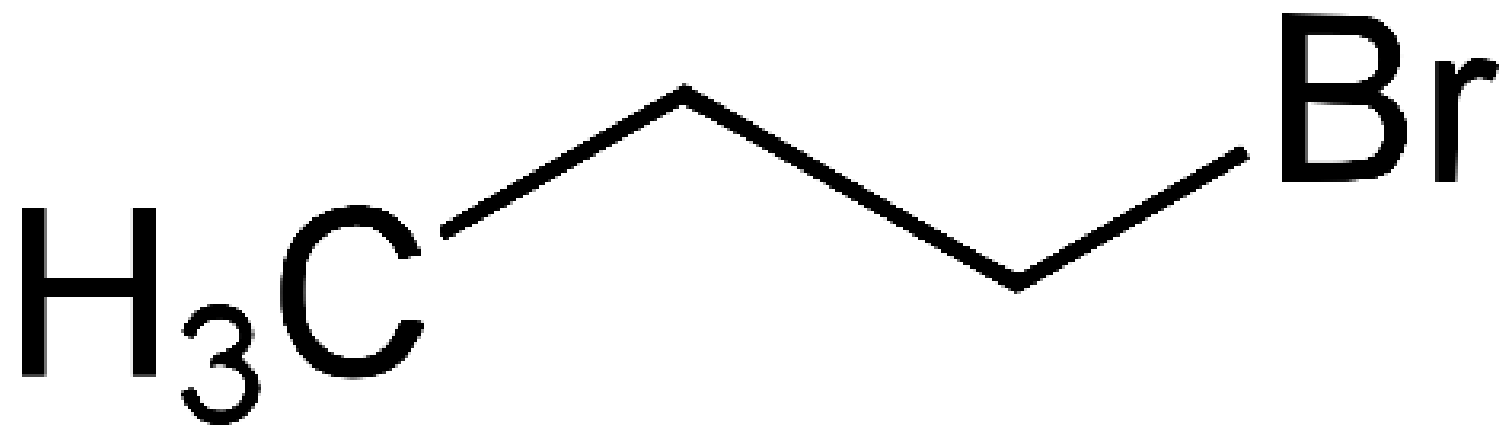
## 1.2 What is methyl octanoate?

- Carboxylic acid ester
- Density: 0,88 g /cm<sup>3</sup>
- **Boiling point: 193 ° C**



# n-propyl bromide

- Bromocarbon
- Density: 1,35 g /cm<sup>3</sup>
- Boiling point: 71 ° C
- ASTM
- Listed as SVHC ( since 4/2020 forbidden in EU)



# Conclusion:

Asphalt extraction and finding binder properties are a task of high complexity

With the use of closed systems, extractions can be performed save and precise

Over the years of analysers development the consumption of solvent dropped down enormously

# Conclusion:

The evaluation process for finding a new solvent is complex

It is not just mixing solvent 1 and 2 together

Extraction is not washing it is testing





# Conclusion:

Knowing the devil and take care of him in the best way with state of the art principles and technologies is smarter then looking for a new one

# Conclusion:

One last question: what are you doing with the dioxins and furans during burning in an ignition oven ?

# Friends for more than 30 years

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## Asphalt

The use of bituminous materials is mainly addressed to road constructions. The asphalt, named also bitumen, is mainly composed by aggregates and binder, with an infinite variation of mixtures. It is therefore necessary to get suitable equipment to perform different test methods and to determine: binder content, internal friction, cohesion, consistency, softening point, viscosity, quality of aggregates, voids percentage, Marshall test and many other parameters.



Lets go.. I wish you all a fantastic 2022

